

7.14

$$E_k = \frac{1}{2} m v^2 = Q = c \cdot m \cdot \Delta t$$

$$100 \text{ km/h} = 100 \frac{10^3 \text{ m}}{3600 \text{ s}} = \frac{100}{3,6} \left[\frac{\text{m}}{\text{s}} \right]$$

$$\frac{1}{2} m v^2 = c \cdot m \cdot \Delta t \quad | : m$$

$$c = 0,45 \cdot 10^3 \left[\frac{\text{J}}{\text{kg} \cdot \text{K}} \right]$$

$$\frac{v^2}{2} = c \cdot \Delta t$$

$$\Delta t = \frac{v^2}{2 \cdot c} = \frac{\left(\frac{100}{3,6} \right)^2}{2 \cdot 0,45 \cdot 10^3}$$

$$\left[\frac{\frac{\text{m}^2}{\text{s}^2}}{\frac{\text{J}}{\text{kg} \cdot \text{K}}} \right]$$

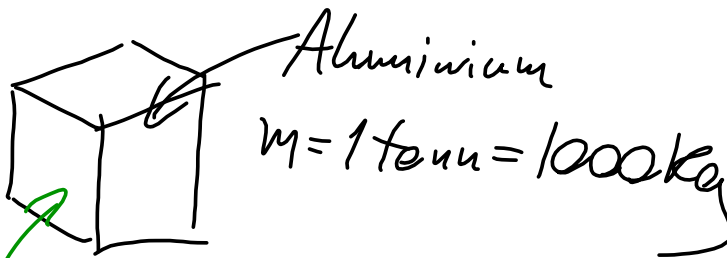
$$\Delta t = \frac{771,6}{0,9 \cdot 10^3}$$

$$\left[\frac{\text{kg} \cdot \frac{\text{m}}{\text{s}^2} \cdot \text{m} \cdot \text{K}}{\text{J}} \right]$$

$$\underbrace{\text{kg} \cdot \frac{\text{m}}{\text{s}^2}}_N \cdot \text{m} = \text{J}$$

$$\Delta t = \frac{771,6}{0,9} \cdot 10^{-3} [\text{K}] = 0,84 [\text{K}]$$

7.18



skal smeltes: fra fast til flytende

Fasevarme: Smeltevarme: $L_s = 395 \frac{\text{kJ}}{\text{kg}}$

$$Q = L_s \cdot m = 395 \cdot 10^3 \left[\frac{\text{kJ}}{\text{kg}} \cdot \text{kg} \right]$$

$$Q = 395 \text{ [MJ]}$$

↑
 må gjøres om til kWh

$$\text{Wh} = \text{J} \quad 1 \text{ h} = 3600 \text{ s}$$

$$\frac{\text{Wh}}{3600} = \text{J}$$

$$\text{s} = \frac{\text{h}}{3600}$$

$$\text{Wh} = 3600 \text{ J} = 3,6 \cdot 10^3 \text{ J}$$

$$1 \text{ kWh} = 10^3 \text{ Wh} = 3,6 \cdot 10^3 \cdot 10^3 \text{ J} = 3,6 \cdot 10^6 \text{ J} = 3,6 \text{ MJ}$$

$$\text{MJ} = \frac{1 \text{ kWh}}{3,6}$$

$$Q = 395 \text{ [MJ]} = 395 \left[\frac{\text{kWh}}{3,6} \right] = 110 \text{ [kWh]}$$

7.20

$$P = 80 \text{ [W]}$$

tid: $\Delta t = (9,5 - 3,5) \text{ min} = 6,0 \text{ min} = 6,0 \cdot 60 \text{ [s]}$
 (leser av kurven)

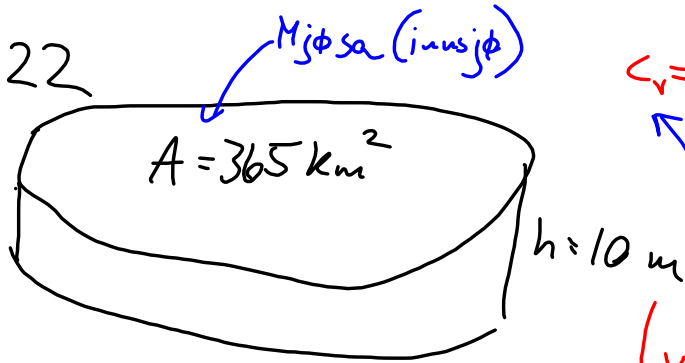
$$= 360 \text{ [s]}$$

$$E = P \cdot \Delta t = 80 \text{ [W]} \cdot 360 \text{ [s]} = 28,8 \cdot 10^3 \text{ [J]} = 28,6 \text{ [kJ]}$$

$$E = Q = L \cdot m$$

$$L = \frac{Q}{m} = \frac{28,6 \text{ [kJ]}}{0,200 \text{ [kg]}} = 144 \frac{\text{[kJ]}}{\text{[kg]}}$$

7.22



$$c_v = 4,2 \frac{\text{kJ}}{\text{kg} \cdot \text{K}}$$

↑ spesifikk varmekapasitet

$$L_v = 334 \frac{\text{kJ}}{\text{kg}}$$

↑ spesifikk smeltekapasitet

a) $\Delta t = 10^\circ \text{C} = 10 \text{ K}$

$$m \Rightarrow V = A \cdot h = 365 \text{ km}^2 \cdot 10 \text{ m} = 365 \cdot 10^6 \text{ m}^2 \cdot 10 \text{ m}$$

$$\text{km}^2 = (10^3 \text{ m})^2 = 10^6 \text{ m}^2$$

$$V = 3650 \cdot 10^6 \text{ m}^3$$

$$1 \text{ m}^3 = 1000 \text{ kg} = 10^3 \text{ kg}$$

(vann)

$$m = 3650 \cdot 10^6 \cdot 10^3 [\text{kg}] = 3,65 \cdot 10^3 \cdot 10^6 \cdot 10^3 [\text{kg}]$$

$$= 3,65 \cdot 10^{12} [\text{kg}]$$

↑ vannet i Mjøsa

$$Q_{10} = c \cdot m \cdot \Delta t = 4,2 \cdot 3,65 \cdot 10^{12} \cdot 10 \left[\frac{\text{kJ}}{\text{kg} \cdot \text{K}} \cdot \text{kg} \cdot \text{K} \right]$$

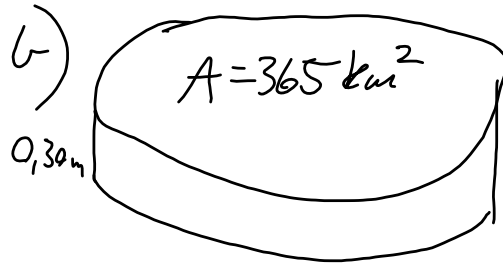
$$Q_{10} = 153,3 \cdot 10^{12} [\text{kJ}]$$

$$1 \text{ kJ} = \frac{1}{3,6} \text{ Wh}$$

$$Q_{10} = 153,3 \cdot 10^{12} \left[\frac{\text{Wh}}{3,6} \right] = 42,58 \cdot 10^{12} [\text{Wh}]$$

$$= 43 [\text{TWh}]$$

7.22



$$m \Rightarrow V = 365 \cdot 10^6 \text{ [m}^2\text{]} \cdot 0,3 \text{ [m]} = 109,5 \cdot 10^6 \text{ [m}^3\text{]}$$

\uparrow
 km^2

$$m = 109,5 \cdot 10^6 \cdot 10^3 \text{ [kg]} = 109,5 \cdot 10^9 \text{ [kg]}$$

\uparrow
 m^3

$$Q = L \cdot m = 334 \left[\frac{\text{kJ}}{\text{kg}} \right] \cdot 109,5 \cdot 10^9 \text{ [kg]}$$

$$= 36,6 \cdot 10^{15} \text{ [J]} = 36,6 \cdot 10^{12} \text{ [kJ]}$$

$$= \frac{36,6}{3,6} \cdot 10^{12} \text{ [Wh]} = 10,2 \text{ [TWh]}$$

c)

$$L_f = 2259 \frac{\text{kJ}}{\text{kg}} \quad \text{5mm}$$

$$5 \text{ mm} \Rightarrow m = 365 \cdot 10^6 \cdot 5 \cdot 10^{-3} \cdot 10^3 \text{ [kg]} = 1825 \cdot 10^6 \text{ [kg]}$$

$$Q = L_f \cdot m = 2259 \left[\frac{\text{kJ}}{\text{kg}} \right] \cdot 1825 \cdot 10^6 \text{ [kg]}$$

$$= 4122675 \text{ [kJ]} = \frac{4122675}{3,6} \text{ [Wh]}$$

$$= 1,1 \text{ [TWh]}$$